

# Oxidative Mechanisms of Toxicity and Susceptibility

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## INTRODUCTION

### Environmental Issue

Oxidation, similar to that encountered in the bleaching of fabric, occurs inside the body and is associated with pollutant exposure and disease:

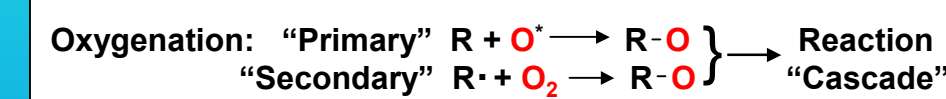
<b>DIRECT OXIDANT POLLUTANTS</b> Particulate matter (peroxides, quinones, metals) Air pollutant gases (ozone, NO <sub>2</sub> , Cl <sub>2</sub> ) Bromate Tobacco smoke	<b>INTERNALLY GENERATED OXIDANTS</b> Injured mitochondria Inflammation: activated phagocytes.
<b>INDIRECT OXIDANTS</b> Metals (cadmium, vanadium, etc.) Asbestos Silica Carbon tetrachloride Dioxin Ultraviolet light Ionizing radiation (radon)	<b>DISEASES</b> Cancer (initiation and promotion) Atherosclerosis Diabetes Asthma Alzheimer's  Inflammatory Disease (asthma, arthritis, autoimmune)

### Study Goals

- 1) To understand the oxidative mode of action of pollutants
- 2) To improve measurements of oxidative stress

### Scientific Approach

"Oxidative stress" involves both electron withdrawal and oxygenation reactions.



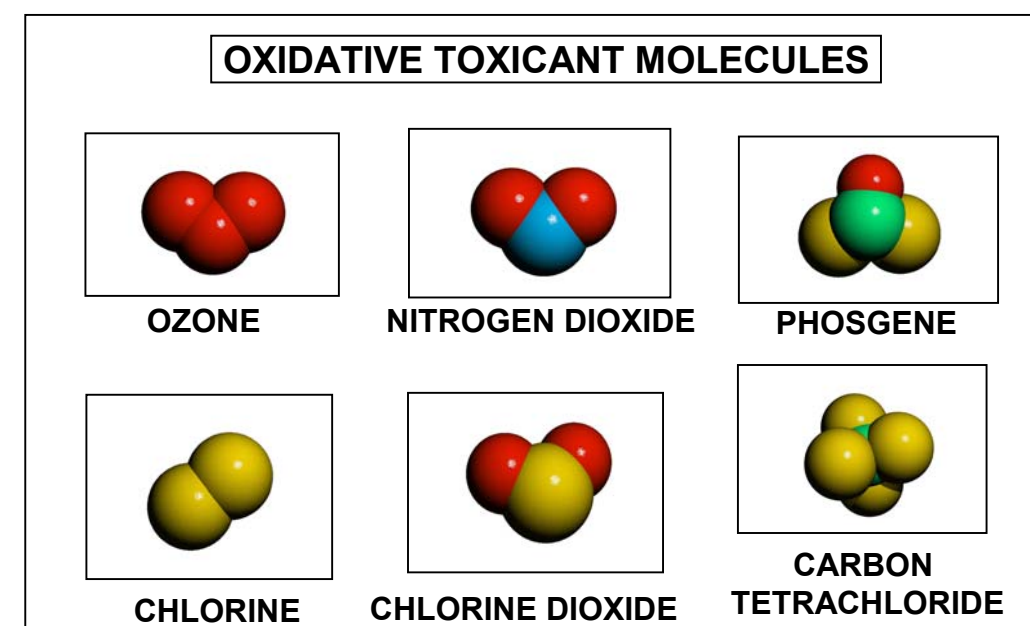
### WHY IS OXYGENATION TOXIC?

It leads to FORMATION OF STRONG, IRREVERSIBLE BONDS

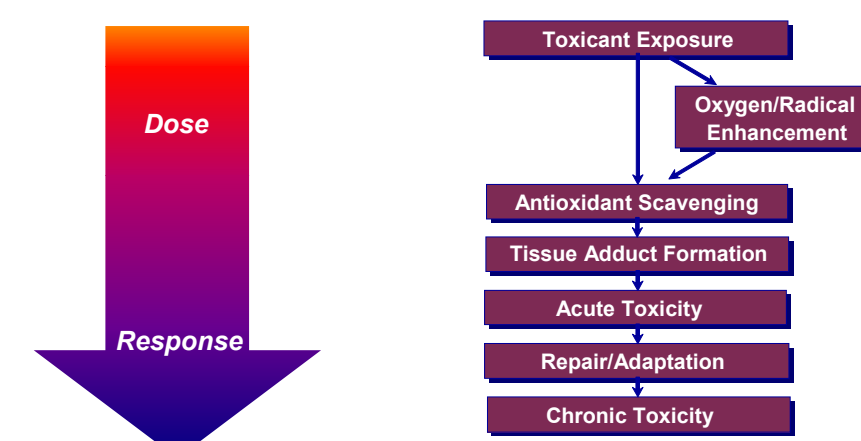
Bond Type	Energy, kcal/mole
C=O	175
C-H	96
C-O	89
C-C	85
S-H	82
C-N	72

## METHODS

### Adduct Formation/Oxygenation



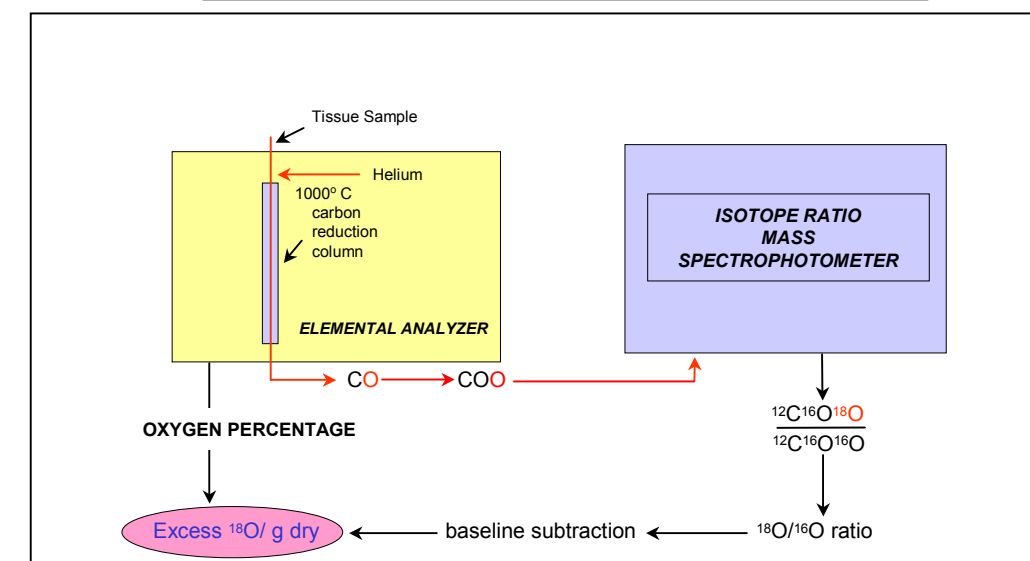
Many toxicants exert their effects through the formation of "tissue adducts" involving oxygen or similar molecules



### Measurement of "oxygenation" reactions

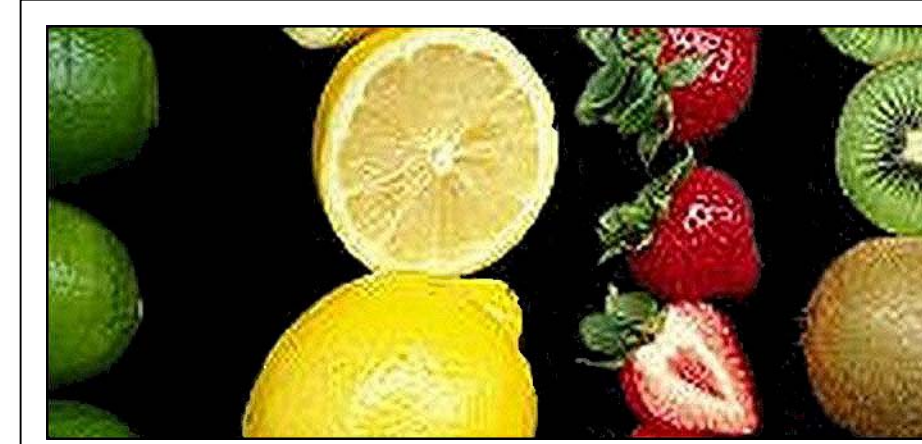
Molecules we have labeled with <sup>18</sup> O		
O <sub>2</sub>	O <sub>3</sub>	H <sub>2</sub> O <sub>2</sub>
OXYGEN	OZONE	HYDROGEN PEROXIDE
ONOO <sup>-</sup>	NO <sub>2</sub>	NO <sup>-</sup>
PEROXYNITRITE ANION	NITROGEN DIOXIDE	NITRIC OXIDE

### MEASUREMENT OF OXYGEN-18 IN TISSUES



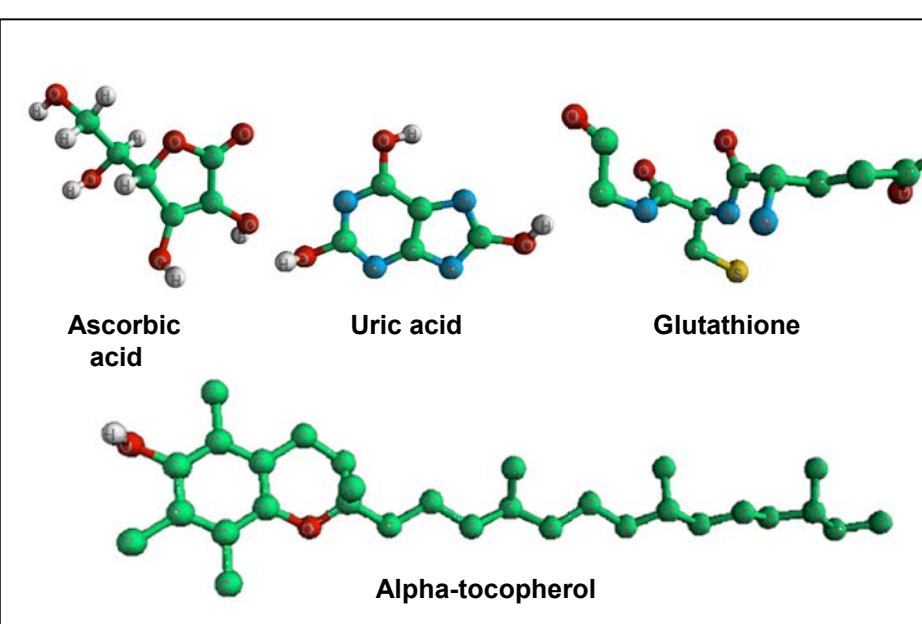
### Antioxidant Measurement

Antioxidants found in fruits and vegetables perform important protective roles in the body.

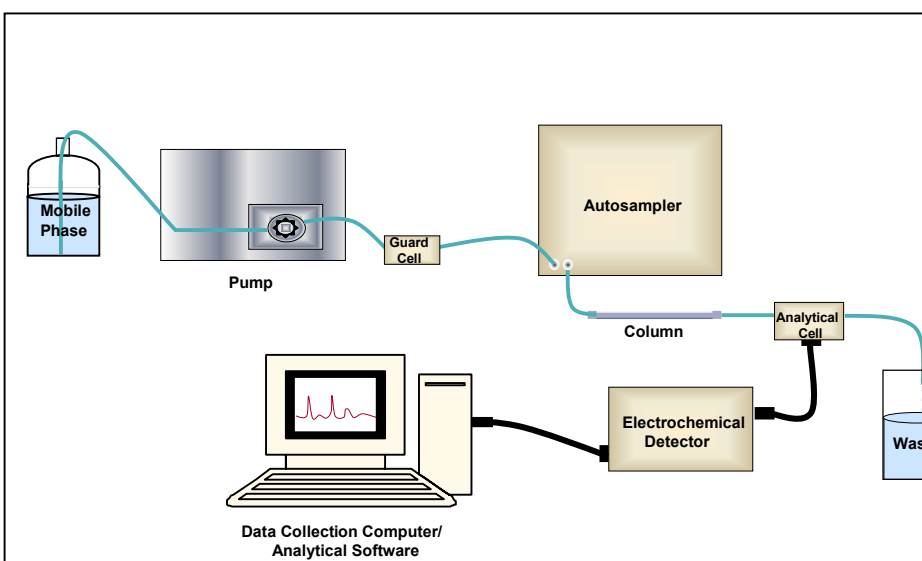


Measurement of antioxidant scavenging is done by measuring the concentration of antioxidant substances

### ANTIOXIDANT SUBSTANCES ASSAYED



### MEASUREMENT OF ANTIOXIDANTS

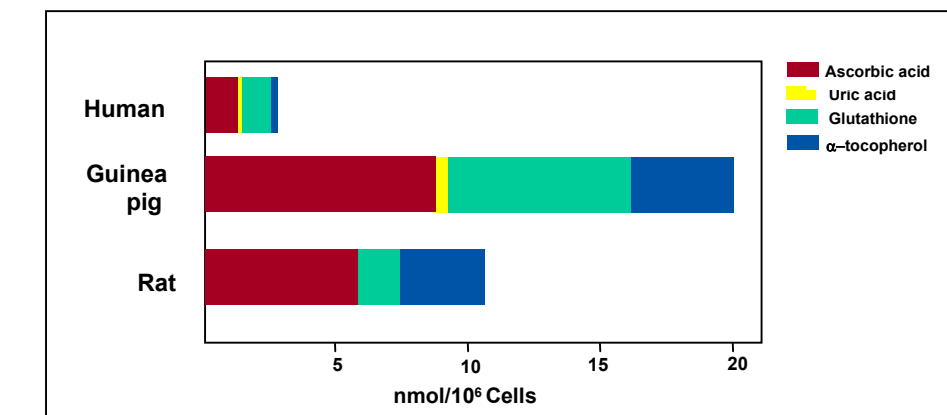


## RESULTS

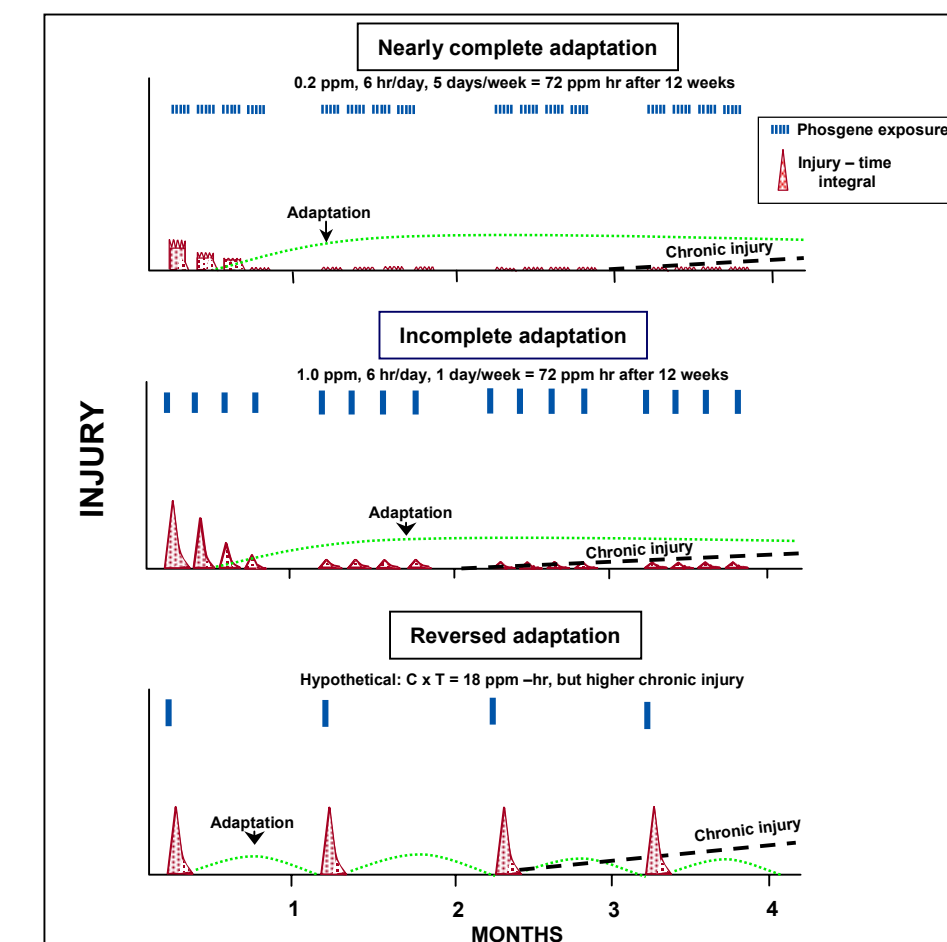
Recent results in Study Goal 1:

To understand the oxidative mode of action.

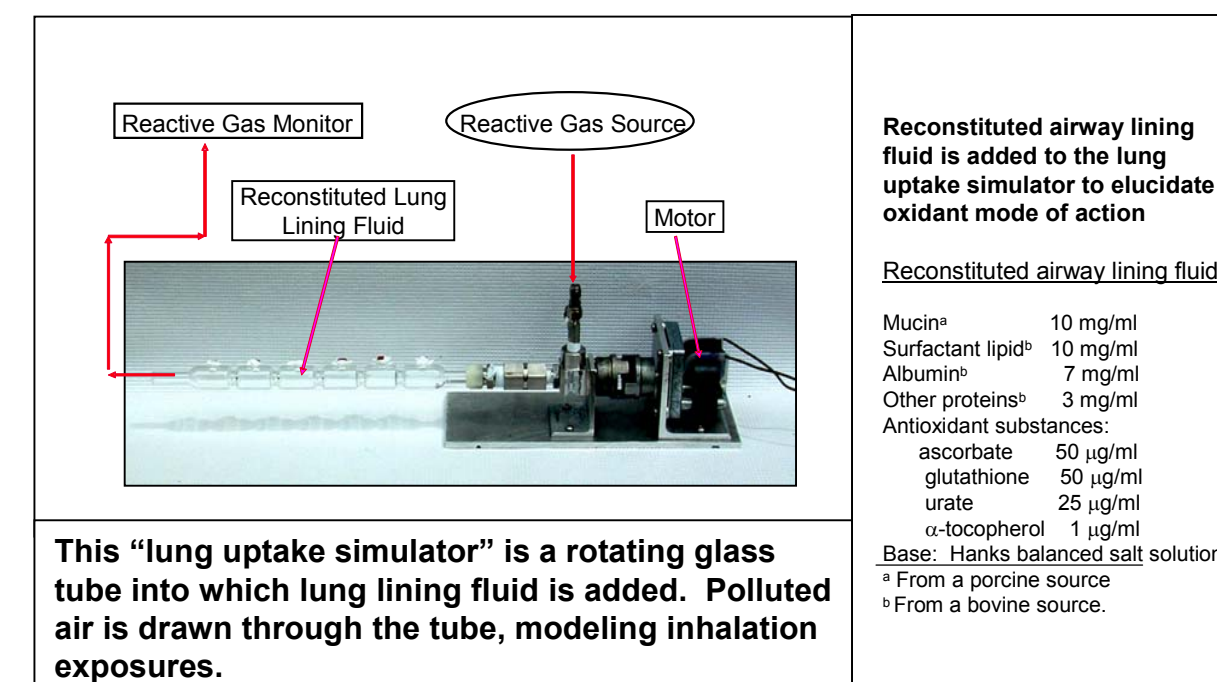
- Humans have low concentrations (compared to rodents) of two important antioxidants: glutathione and ascorbate. <sup>1</sup>



- Rodents depleted of antioxidants to the level of humans have increased susceptibility to both ozone and PM. <sup>2</sup>
- Ozone dosimetry studies suggest that 1) species with high oxygen consumption are most susceptible, and 2) these species defend against injury by lowering activity and body temperature.
- Progression of acute injury to toxic injury appears to be most severe following episodic exposures rather than continuous exposures. <sup>3</sup>



- A "lung uptake simulator" has been developed which allows direct modeling of oxidative interactions in airway lining fluids.

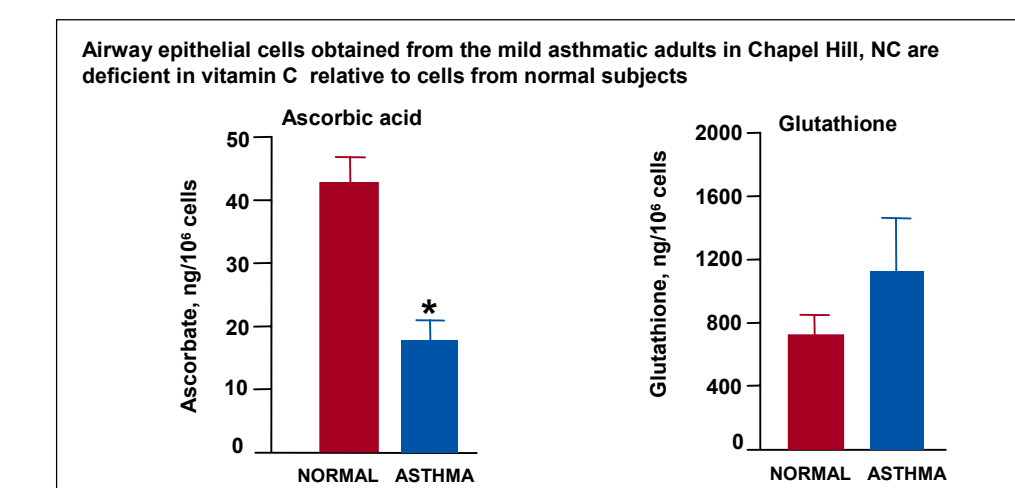


This "lung uptake simulator" is a rotating glass tube into which lung lining fluid is added. Polluted air is drawn through the tube, modeling inhalation exposures.

Recent results in Study Goal 2:

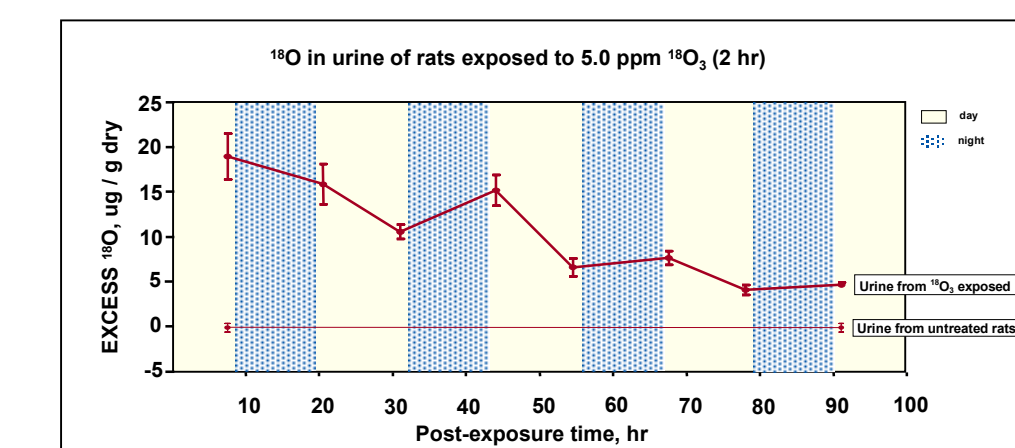
To improve measurements of oxidative stress.

- Oxidative stress in mild asthmatics has been measured through a decrease in ascorbate concentrations in induced sputum. <sup>5</sup>

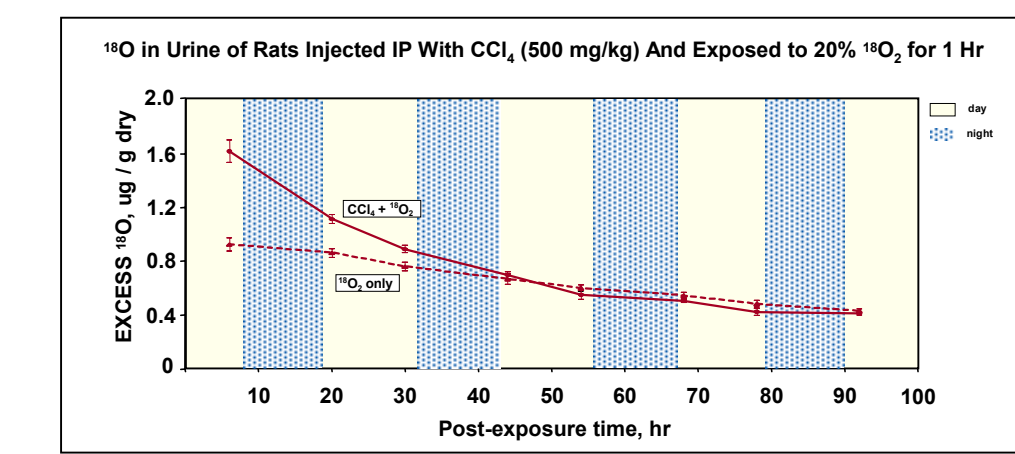


- Oxidative stress in asthmatic children was shown to be ameliorated through antioxidant supplements. <sup>6</sup>
- Non-invasive biomarkers of oxidative stress for use in human epidemiology have been identified in studies with rats:

- Carbon tetrachloride induced oxidative stress to the liver was detectable in blood plasma and urine through measurements of isoprostanes, malondialdehyde adducts and GSSG/GSH ratios. <sup>7</sup>
- Most blood plasma antioxidants were not good biomarkers due to interferences caused by liver spillage of antioxidants. <sup>7</sup>
- Liver cell rupture assays measured injury more sensitively than direct oxidant adduct measurements. <sup>7</sup>
- Ozone induced injury to the lung was not detectable by any blood plasma or urine measurement that was effective for carbon tetrachloride, however.
- Oxygen-18 labeling made possible detection of excreted ozone adducts in the urine.
- Oxygen-18 labeling also indicated that ozone adducts are not recycled by the body, but are completely excreted within about four days.



- The use of oxygen-18 appears to make possible comparison of total-body oxidative stress between carbon tetrachloride and ozone, and also suggest the potential for finding even better non-invasive biomarkers.



### IMPACTS:

- Finding biomarkers for oxidative stress will assist in future epidemiology studies.
- Quantification of oxidative stress will enable scientific extrapolation of animal toxicity data to humans.
- A better understanding of the role of antioxidants in modifying injury will enable improved risk assessments and better protection of susceptible individuals.
- In vitro model systems may reduce the number of experimental animals needed for toxicity studies.

### FUTURE STUDIES

- Dose response modeling of chlorine gas for an Air Toxics studies involving human, animal and in vitro comparisons.
- Dosimetry and toxicity assessment of chlorine dioxide gas as part of a new Homeland Security study (see other poster).
- Dosimetry and toxicity assessment of "sentinel" species for Homeland Security (see other poster).
- Dosimetry and toxicity study of ingested bromate for the Water Program.
- Continuation of the Biomarkers of Oxidative Stress Study using endotoxin as the oxidative stressor.
- Two goal 8 studies examining blood plasma and tissue concentrations of antioxidant substances.
- External collaborations involving antioxidant pharmacokinetics and age susceptibility to ozone.

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